Carbon Availability and Nitrogen Mineralization Control Denitrification Rates and Product Stoichiometry during Initial Maize Litter Decomposition

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Abstract: Returning crop residues to agricultural fields can accelerate nutrient turnover and increase N_2O and NO emissions. Increased microbial respiration may lead to formation of local hotspots with anoxic or microoxic conditions promoting denitrification. To investigate the effect of litter quality on CO_2, NO, N_2O, and N_2 emissions, we conducted a laboratory incubation study in a controlled atmosphere (He/O_2, or pure He) with different maize litter types (*Zea mays* L., young leaves and roots, straw). We applied the N_2O isotopocule mapping approach to distinguish between N_2O emitting processes and partitioned the CO_2 efflux into litter- and soil organic matter (SOM)-derived CO_2 based on the natural ^{13}C isotope abundances. Maize litter increased total and SOM derived CO_2 emissions leading to a positive priming effect. Although C turnover was high, NO and N_2O fluxes were low under oxic conditions as high O_2 diffusivity limited denitrification. In the first week, nitrification contributed to NO emissions, which increased with increasing net N mineralization. Isotopocule mapping indicated that bacterial processes dominated N_2O formation in litter-amended soil in the beginning of the incubation experiment with a subsequent shift towards fungal denitrification. With onset of anoxic incubation conditions after 47 days, N fluxes strongly increased, and heterotrophic bacterial denitrification became the main source of N_2O. The $N_2O/(N_2O+N_2)$ ratio decreased with increasing litter C:N ratio and $C_{org}:NO_3^-$ ratio in soil, confirming that the ratio of available C:N is a major control of denitrification product stoichiometry.

Keywords: fungal denitrification; nitrification; isotopocules; priming effect; nitric oxide; nitrous oxide; dinitrogen; greenhouse gas; decomposition

1. Introduction

Returning of crop residues is a common agricultural management strategy to prevent nutrient losses and to increase soil fertility. However, acceleration of N and C cycling processes often lead to increased losses of climate-relevant gases.

Addition of plant litter to soils has been proven to increase CO_2 and N_2O emissions over a vast range of soil conditions and litter types [1–4]. Upon degradation, plant litter provides nutrients for decomposing and denitrifying microorganisms. Thus, variations in N_2O emissions have often been related to litter quality, especially the C:N ratio [1,2]. For litter with C:N < 25:1, mineralization increases soil NO_3^- content leading to increased denitrification [5,6], while for C:N > 25:1, N is immobilized by soil microorganisms to decompose litter C compounds [7] and restricts N_2O emissions [8]. When litter quality

was analyzed in more detail, easily degradable fractions explained a large share of the variability of N_2O emissions [9,10], while the lignin content was not relevant [11,12]. Recent studies confirm that the quality of C compounds (especially water-soluble C) from litter is a main driver of denitrification after litter addition [13,14].

Easily degradable C compounds (e.g., sugars, proteins, amino acids, and carbohydrates) control litter decomposition dynamics in the initial phase and subsequent CO_2 efflux from soils [4,15,16]. Furthermore, the quality of organic substrates affects decomposition of soil organic matter (SOM) [17]. Readily accessible high-quality substrates increase SOM decomposition, leading to a positive priming effect in soils [18,19]. When litter and SOM turnover are increased after litter addition, microbial O_2 demand increases with increasing microbial respiration. This may lead to formation of local hotspots with anoxic or microoxic conditions providing favorable conditions for denitrifying soil microorganisms [20]. Accordingly, a recent study reported the highest denitrification-derived N_2O losses in soils with high SOM priming after addition of labile C substrates (glucose, vanillin) [21]. However, further studies with plant residues are necessary to better understand the interactions between C turnover and denitrification.

The aim of this study was to investigate the effect of litter C quality and SOM turnover on denitrification. We anticipate that increased SOM turnover after litter addition promotes the formation of anoxic hotspots for denitrification and expect higher litter and SOM turnover from litter with high degradability. Thus, we hypothesize that (i) N_2O fluxes from denitrification increase when C turnover from litter and SOM is high, leading to (ii) higher N_2O+N_2 losses when litter with a high share of easily degradable C is added, while (iii) the $N_2O/(N_2O+N_2)$ ratio is controlled by the availability of C_{org} in relation to NO_3^-.

Therefore, we setup a laboratory incubation experiment in an artificial N_2-free atmosphere under fully controlled conditions. We compared different types of maize litter (fresh leaves and roots, straw) and investigated the effect of litter quality on total CO_2, NO, N_2O, and N_2 emissions. To trace maize litter (C4 plant) and SOM turnover, we used a grassland soil whose organic C originates solely from C3 vegetation and partitioned the CO_2 efflux into its sources (i.e., litter- and SOM-derived CO_2) based on the natural ^{13}C isotope abundances. In addition, we analyzed the intramolecular distribution of the naturally occurring ^{15}N and ^{18}O isotopes in the linear N_2O molecule and applied the N_2O isotopocule mapping approach to estimate the contribution of denitrification to N_2O formation [22,23].

2. Materials and Methods

2.1. Preparation of Soil and Plant Material

The soil for the experiment was taken from a long-term field experiment at the grassland research station of the University of Gießen (latitude N50°32′, longitude E8°41.3′, elevation 172 m a.s.l.), sieved to 10 mm, air-dried, and stored at 4 °C. The soil was classified as Fluvic Gleysol of clay loam texture (32% clay, 41% silt, and 27% sand) with a pH ($CaCl_2$) of 5.67. Total soil N content was 0.42%, total soil C content was 4.2%, and $\delta\ ^{13}C$ was $-28.37‰$. Prior to the incubation experiment, the soil was pre-incubated in the dark for 5 weeks at 50% water holding capacity (WHC) and 20 °C.

Maize plants (*Zea mays* L. cv. Ronaldinio) were grown in nutrient solution [24] for 5 weeks. Leaves were cut from stems and left to wilt at room temperature for 4 h. Roots were rinsed with H_2O_{dest} and carefully dried with paper towels. Leaves and roots were stored at 4 °C and 90% relative humidity until experimental setup. Maize straw was collected from an experimental field site of the University of Göttingen after grain harvest in October 2018. Maize straw was shock frozen in liquid N_2 and stored at -20 °C. Prior to setting up the experiment, maize straw was unfrozen and all maize litter was cut to a size of 2 cm. A subsample of soil and maize litter was analyzed for total N, total C, and $\delta\ ^{13}C$ using an elemental analyzer (NA1110, CE-Instruments, Rodano, Milano, Italy) linked to a gas-isotope ratio mass spectrometer (Delta Plus, Finnigan MAT, Bremen, Germany) via a Conflo III Interface (Finnigan MAT, Bremen, Germany). Further, plant litter was

analyzed for water-extractable C and N content. Briefly, 0.2 g of finely ground plant litter were extracted in 100 mL H_2O_{bidest}, shaken for 1 h, filtered with 0.45 µm polyether sulfone filters (Labsolute, Renningen, Germany), and stored at $-20\ °C$. The extracts were analyzed for organic C and total N content using a multi N/C® Analyzer (Analytik Jena, Jena, Germany). Another subsample of finely ground plant litter was analyzed by ^{13}C solid-state cross polarization magic angle spinning nuclear magnetic resonance spectroscopy (^{13}C-CPMAS NMR) using a Bruker Avance^{III} 200 spectrometer (Bruker BioSpin GmbH, Karlsruhe, Germany). Samples were weighed into zircon oxide rotors and spun around a magic angle at a speed of 6.8 kHz. Contact time was 1 ms and the recycle delay time was set to 2 s, line broadening was set at 0. Peak integration areas were separated into -10–45 ppm (alkyl C), 45–110 ppm (O/N-alkyl C), 110–160 ppm (aryl C), and 160–220 ppm (carboxylic C).

2.2. Automatized Laboratory Incubation Experiment and Gas Analysis

The incubation experiment was carried out under fully controlled conditions using an automated soil incubation system with artificial atmosphere similar to systems described earlier [25–28]. For the incubation experiment, soil moisture was adjusted to 70% water-filled pore space (WFPS, equivalent to 67.7% WHC or 31.9% gravimetric water content) and 50 mg N kg^{-1} was added by spraying a KNO_3 solution onto the soil and thoroughly stirring it with a spoon. For treatments with litter, litter was homogenously mixed with soil (Maize leaves: 40.5 g FM kg^{-1}, maize roots: 42 g FM kg^{-1}, maize straw: 12.8 g FM kg^{-1}). The soil for each pot was prepared separately to ensure the same amount of litter was added. Then, the equivalent of 2.5 kg dry soil was filled into acrylic glass pots (inner diameter 172 mm, height 210 mm) with a porous ceramic plate at the bottom and compacted in a stepwise mode by filling a 2 cm-layer of soil in pots and compacting it with a plunger. To ensure continuity between soil layers, the surface of the compacted layer was gently scratched before adding the next soil layer. Soil height in the pots was 10 cm, and bulk density was 1.1 g cm^{-3}. Each litter treatment was replicated five times, a control treatment without litter was replicated four times, and one empty pot was included as reference to determine background gas concentrations.

Pots were tightly closed with transparent acrylic glass lids with rubber seals, and the outside of the pots was covered with dark plastic sheets to prevent algae growth. Pots were alternately evacuated using a rotary vacuum pump (Pfeiffer Vacuum GmbH, Asslar, Germany) and flushed with a gas mixture (80% He, 20% O_2) for 24 h. The gas mixture was prepared by using stainless steel capillaries of different length and inner diameter. For the first cycles, pots were evacuated from the top and the bottom and, subsequently, flushed with the He/O_2 gas mixture. Then, pots were evacuated from the bottom and simultaneously flushed from the top to replace the atmosphere inside the soil column. For measurements, the outlet of all pots was connected to flow-through multi-position valves (16 ports, Vici Valco Instruments, Houston, TX, USA) with multi-position actuator control modules (Vici Valco Instruments, Houston, TX, USA), and controlled by Trilution Software (Gilson Inc., Middleton, WI, USA) via an interface module (506C System Interface, Gilson Inc., Middleton, WI, USA). The selected stream outlet tube of the multi-position valve was connected to a gas chromatograph (GC-450, Bruker, Billerica, USA) equipped with a thermal conductivity detector (TCD) for measurement of CO_2 and a pulsed discharge detector (PDD, Vici AG International, Schenkon, Switzerland) for N_2O and N_2. The sample gas outlet of the GC was connected to a flow-through massflowmeter (Alicat Scientific, Tucson, AZ, USA), and a trace-level gas analyzer (CLD 88Yp, Eco Physics AG, Dürnten, Switzerland) equipped with a chemoluminescence detector (CLD) to analyze NO concentrations. To add up to the required 300-mL-flow of the NO analyzer, samples were diluted with synthetic air. Processing of GC data was done using CompassCDS software (SCION Instruments, Livingston, UK). Data from the NO analyzer and flowmeter were read out every 10 s via a serial port.

The analytical precision of the GC was determined by repeated measurements of standard gases (CO_2, N_2O, N_2) and was consistently < 2%. Detection limits were 0.08 µg N_2O-N kg^{-1} h^{-1} and 5.5 µg N_2-N kg^{-1} h^{-1}. The non-selected outlet streams of the multi-position valves were used to sample headspace gas for analysis of isotopic compositions ($\delta^{13}C$ of CO_2, isotopocules of N_2O). After 47 days, the pots were flushed with pure helium to establish anoxic conditions to determine potential denitrification. After 8 days of anoxic incubation (55 days in total), the pots were opened for final sampling.

2.3. $^{13}CO_2$ Sampling, Analysis, and Calculations

For determination of $\delta^{13}C$ of soil-emitted CO_2, samples were flushed into 12 mL Exetainer® septum-capped vials (Labco, High Wycombe, UK). Samples were taken twice a day for the first 5 days, daily for the next 12 days, every second day for the next 14 days, and every 3 days until day 43. Samples were introduced by a Combi-Pal autosampler (CTC-Analytics, Zwingen, Switzerland) to a GC (GC-Box, Thermo Fisher Scientific, Bremen, Germany) coupled to an isotope ratio mass spectrometer (Delta plus XP, Thermo Fisher Scientific, Bremen, Germany) via a Conflo III Interface (Thermo Fisher Scientific, Bremen, Germany). The fractions of CO_2 derived from litter (f_{litter}) and SOM (f_{SOM}) were calculated using Equations (1) and (2):

$$f_{litter} = (\delta^{13}C_{treatment} - \delta^{13}C_{Control})/(\delta^{13}C_{litter} - \delta^{13}C_{Control}) \tag{1}$$

$$f_{litter} + f_{SOM} = 1 \tag{2}$$

where $\delta^{13}C_{treatment}$ is the measured $\delta^{13}C$ (‰) of CO_2 from litter treatment, $\delta^{13}C_{Control}$ is the measured $\delta^{13}C$ (‰) of CO_2 from control treatment without litter addition, and $\delta^{13}C_{litter}$ is the mean measured $\delta^{13}C$ (‰) of CO_2 lost from maize litter (Leaf: -7.91 δ‰, Root: -7.50 δ‰, Straw: -9.33 δ‰, see supplement for details, Figure S2). For each treatment, the priming effect (*PE*) was calculated as the difference between SOM-derived CO_2-C (C_{SOM}) and CO_2-C from control treatment without litter ($C_{Control}$) (3):

$$PE = C_{SOM} - C_{Control} \tag{3}$$

2.4. $^{15}N_2O$ Sampling, Analysis, and Isotopocule Mapping Approach

On 2, 5, 8, 15, 23, 31, 39, 48, 51, and 54 days after onset of incubation (DAO), samples for analyses of N_2O isotopomers were flushed into 100 mL crimp-top vials with butyl rubber septa. Samples were analyzed on a gas-isotope ratio mass spectrometer (Delta plus XP, Finnigan MAT, Bremen, Germany) coupled to a trace gas pre-concentration unit Precon (Thermo Electron Cooperation, Bremen, Germany) via a GC/GP Interface (Thermo Electron Cooperation, Bremen, Germany). In this setup, m/z 44, 45, and 46 of the intact N_2O^+ molecular ions and m/z 30 and 31 of the NO^+ fragment ions are measured simultaneously [29], and $\delta^{15}N^{bulk}_{N2O}$, $\delta^{15}N^{\alpha}_{N2O}$, and $\delta^{18}O_{N2O}$ values were determined [30]. $\delta^{15}N^{\beta}_{N2O}$ values were calculated based on the following Equation (4):

$$\delta^{15}N^{bulk}_{N2O} = (\delta^{15}N^{\alpha}_{N2O} + \delta^{15}N^{\beta}_{N2O})/2 \tag{4}$$

Site preference ($\delta^{15}N^{SP}_{N2O}$) was calculated as the difference between $\delta^{15}N^{\alpha}_{N2O}$ and $\delta^{15}N^{\beta}_{N2O}$. We used the scrambling factor of 0.096 determined by Buchen et al. (2018) [31] to correct measured data [32]. $\delta^{18}O$ of soil water was -6.7 δ‰.

We applied the isotopocule mapping approach by Lewicka-Szczebak et al. (2017) [22] to calculate the fraction of residual unreduced N_2O (rN_2O) and the N_2O fraction from heterotrophic bacterial denitrification (f_{bD}) based on the sample position in the $\delta^{15}N^{SP}/\delta^{18}O$ map using a mixing equation for the bacterial fraction (6) and the Rayleigh equation for N_2O reduction (5):

$$r_{N2O} = e^{((\delta r - \delta 0)/\eta red)} \tag{5}$$

$$\delta_{0\_sample} = \delta_{bD} * f_{bD} + \delta_{fD/Ni} * (1 - f_{bD}) \tag{6}$$

where δ_r is the isotopic signature of residual N_2O after partial reduction, δ_0 is the isotopic signature of initial N_2O before reduction, and η_{red} is the isotopic fractionation factor associated with N_2O reduction to N_2. Two main scenarios were considered: (1) N_2O emitted from bacterial denitrification is first reduced to N_2 and residual N_2O is then mixed with N_2O originating from nitrification or fungal denitrification (Scenario 1, reduction-mixing). Alternatively, (2) N_2O from bacterial denitrification and nitrification or fungal denitrification is first mixed and then partially reduced to N_2 (Scenario 2, mixing-reduction). Recently, non-overlapping signatures for N_2O produced by nitrification or fungal denitrification were proposed [23], and we calculated both scenarios for mixing of bacterial denitrification with either nitrification or fungal denitrification. A detailed description of the calculations can be found in [33]. We used the isotopic fractionation factors proposed by [23] (Supplementary Table S1), which were corrected for $\delta^{18}O$ of soil water ($-6.7\,\delta‰$) for mapping and calculations.

In addition, calculations can be based on minimum or maximum end-member values, fractionation factors, and reduction factors, leading to a total of 14 different scenarios [31]. In our study, we used mean values for mixing, fractionation, and reduction whenever possible. However, as during anoxic incubation, samples were distributed outside the mean mixing-reduction area, we used minimum reduction values (mean mixing, mean fractionation) for 51 and 54 DAO. When calculations yielded values < 0 or > 1 for fbD or rN_2O, these values were removed from the dataset before calculating means and plotting.

2.5. Soil Analyses

Samples of pre-incubated soil were taken prior to experimental setup. After opening pots at the end of the experiment, soil from each pot was homogenized and a sample was taken for analyses. Subsamples were analyzed for soil mineral N, water-extractable organic C (WEOC), and soil water content. For analysis of mineral N (NO_3^- and NH_4^+), 50 g fresh soil were weighed into plastic bottles and frozen at $-20\,°C$ until analysis. Frozen samples were extracted with 2 *M* KCl solution (1:5 *w:v*) and shaken on an overhead shaker for 60 min. Samples were filtered with 615 $\frac{1}{4}$ filter paper (Macherey–Nagel GmbH & Co. KG, Düren, Germany) and extracts were analyzed colorimetrically using the San^{++}Continuous-Flow Analyzer (Skalar Analytical B.V., Breda, The Netherlands). To determine isotopic signatures of nitrate, the bacterial denitrification method with *Pseudomonas aureofaciens* was applied [34,35].

WEOC was extracted by homogenizing 2 g of fresh soil with 10 mL H_2O_{bidest}. Samples were centrifuged, filtered with 0.45 μm polyether sulfone filters (Labsolute, Renningen, Germany), and stored at $-20\,°C$. The extracts were analyzed for organic C and total N content using a multi N/C$^®$ Analyzer (Analytik Jena, Jena, Germany). Soil water content was determined by oven drying at $105\,°C$.

2.6. Calculations and Statistics

For all calculations and statistical analyses, the statistical software R version 3.6.0 [36] was used. Fluxes of CO_2, N_2O, N_2, and NO (F, μg kg^{-1} h^{-1}) were calculated using the dynamic chamber approach (7):

$$F = (C_o - C_i) * Q/m \tag{7}$$

where C_o is the concentration at the outflow and C_i is the concentration at the inflow of each vessel (mg N m^{-3}, or mg C m^{-3}), Q is the flow rate through the headspace (m^3 h^{-1}), and *m* is the dry mass of soil per vessel (kg).

Net N mineralization was calculated according to Equation (8):

$$\text{Net N mineralization} = NO_3^-{}_{end} + NH_4^+{}_{end} + NO_{cml} + N_2O_{cml} + N_2{}_{cml} - (NO_3^-{}_{start} + NH_4^+{}_{start} + NO_3^-{}_{fertilizer}) \tag{8}$$

Mean values and standard deviations were calculated using the *SlidingWindow* function from the package *evobiR* v.1.1 [37] or the *rollapply* function from the package *zoo* [38].

Cumulative emissions were calculated by interpolation between measured fluxes. To test for differences between treatments, a one-way ANOVA was calculated when data were normally distributed or the Kruskal–Wallis rank sum test for non-normally distributed data followed by the LSD post hoc test. A *t*-test at $p < 0.05$ was used to compare soil NO_3^-, NH_4^+, and WEOC content before and after the incubation. To analyze the effect of litter input and litter quality on CO_2 and N emissions, simple linear regression models were tested. In all plots, color schemes from the R package viridisLite v0.3.0 [39] were used.

3. Results

3.1. Characterization of Maize Litter

Maize litter types differed in their chemical composition (Table 1). Total C content ranged between 40% in maize roots and 47% in maize leaves. Total N content ranged between 3.8% in maize leaves and 0.85% in maize straw. C:N ratio was highest in maize straw (51.4) and similar in maize leaves and roots (12.3 and 13.8, respectively). Water-soluble C contents were similar in all maize litter types (8–8.5%). Water-soluble N content was highest in maize roots (1.22%) and lowest in maize straw (0.39%). Thus, water-soluble C:N was highest in maize straw and lowest in maize roots. ^{13}C-CPMAS NMR spectroscopy of maize litter revealed that maize straw and maize leaves were closer in their chemical composition than maize roots (Table 1, spectra in Supplementary Figure S1). Maize roots were characterized by the lowest shares of alkyl C and carboxyl C and the highest share of O/N-alkyl C, while maize leaves had highest shares of carboxyl C and alkyl C.

3.2. Soil N and C Content

Soil NO_3^- content increased in Control, Leaf, and Root treatments during the incubation experiment due to a net mineralization of N (Table 2). In contrast, addition of maize straw significantly decreased soil NO_3^- content and immobilized N. Soil NH_4^+ content strongly decreased in all treatments during the incubation period and was significantly higher in all maize litter treatments than in Control, but differences between treatments were small. WEOC content increased in all maize litter treatments, but did not change in Control. No differences were found in soil WEOC content between different litter treatments at the end of the experiment.

$\delta^{15}N_{NO3}$ and $\delta^{18}O_{NO3}$ of added KNO_3 were higher compared to initial soil NO_3^- at onset of incubation (Table 2). At the end of the incubation experiment, $\delta^{15}N_{NO3}$ and $\delta^{18}O_{NO3}$ of soil NO_3^- differed between litter treatments. The lowest $\delta^{15}N_{NO3}$ was measured in Root, and the lowest $\delta^{18}O_{NO3}$ in Control and Leaf. The highest $\delta^{15}N_{NO3}$ and $\delta^{18}O_{NO3}$ were measured in Straw. $\delta^{15}N_{NO3}$ increased with decreasing net N mineralization (adj. $R^2 = 0.73$, $p < 0.001$, Table 3).

Table 1. Characteristics of maize litter used in the incubation experiment: total and water-soluble C and N content, C:N ratios, $\delta\ ^{13}$C, and distribution of C species in different plant litter types (values represent the percentage contribution of the different integrated chemical shift regions determined by ^{13}C-CPMAS NMR spectroscopy).

	Dry Matter (%)	C (%)	$\delta\ ^{13}$C (‰)	N (%)	C:N	Water Soluble C (%)	Water Soluble N (%)	Water Soluble C:N	Alkyl C (%)	O/N-Alkyl C (%)	Aryl C (%)	Carboxyl C (%)
Maize leaves	27.9	46.58	−14.70	3.80	12.27	8.03	0.69	11.6	16.05	63.67	10.96	9.31
Maize roots	7.8	40.12	−12.97	2.90	13.82	8.53	1.22	7.0	8.08	80.65	10.19	1.10
Maize straw	31.4	43.84	−14.11	0.85	51.36	8.25	0.39	21.4	11.48	69.93	11.23	7.23

Table 2. Soil mineral N and water-extractable organic C (WEOC) before setup (initial) and at the end of the incubation experiment. Net N mineralization over the whole incubation period of 55 days.

	NO_3^- (mg N kg^{-1} dry soil)		NH_4^+ (mg N kg^{-1} dry soil)		WEOC (mg C kg^{-1} dry soil)		Net N Mineralization (mg N kg^{-1} dry soil)		$\delta^{15}N_{NO3}$ of Soil NO_3^- (‰)		$\delta^{18}O_{NO3}$ of Soil NO_3^- (‰)	
Initial	102.9 ± 4.59		34.7 ± 3.16		51.4 ± 6.49		-		−5.74 ± 0.19/2.44 ± 0.22 [1]		1.84 ± 0.29/22.95 ± 0.40 [1]	
Control	142.1 ± 7.8	b ***	4.12 ± 0.27	c ***	56.3 ± 8.0	b	26.4 ± 5.5	b	7.80 ± 0.58	bc	9.21 ± 1.05	c
Maize Leaves	169.6 ± 4.4	a ***	5.84 ± 0.28	a ***	76.4 ± 3.1	a ***	75.5 ± 15.1	a	8.58 ± 0.24	b	9.27 ± 0.31	c
Maize Roots	176.1 ± 6.9	a ***	4.85 ± 0.39	b ***	72.0 ± 5.7	a ***	69.3 ± 5.4	a	6.64 ± 0.62	c	11.49 ± 0.52	b
Maize Straw	70.6 ± 5.6	c ***	6.00 ± 0.60	a ***	71.9 ± 5.3	a ***	−26.8 ± 5.7	c	16.40 ± 1.63	a	14.53 ± 0.34	a

Values represent means ± standard deviation (n = 5, Control and Initial n = 4). Different letters in the same column indicate a significant difference according to the LSD post hoc test at $p \le 0.05$. *** indicates a significant difference to Initial content according to *t*-test at $p \le 0.05$. [1] initial soil NO_3^- after pre-incubation/added KNO_3 (means ± standard deviation, n = 4).

Table 3. Coefficients of determination and *p*-values for simple linear regressions.

	Adjusted R^2	*p*-Value	n
Oxic incubation period			
Cumulative NO+N_2O emissions ~ water-soluble litter C input	0.4401	0.001172	19
Cumulative N_2O emissions ~ litter C:N ratio	0.247	0.03428	15
Cumulative NO emissions ~ water-soluble litter C:N ratio	0.8703	2.427×10^{-7}	15
Cumulative NO emission ~ net N mineralization	0.5671	0.0001197	19
NO+N_2O flux ~ CO_2 flux	0.08023	$<2.2 \times 10^{-16}$	1715
Anoxic incubation period			
Cumulative N_2 emissions ~ water-soluble litter C:N ratio	0.2553	0.03158	15
Cumulative NO+N_2O+N_2 emissions ~ total litter C input	0.5087	0.0003655	19
N_2O/(N_2O+N_2) ratio ~ water-soluble litter C:N ratio	0.5061	1.886×10^{-6}	19
N_2O/(N_2O+N_2) ratio ~ WEOC: NO_3^- ratio	0.4127	0.0018	19
NO+N_2O+N_2 flux ~ CO_2 flux	0.864	$<2.2 \times 10^{-16}$	176
Total incubation period			
Cumulative CO_2 emissions ~ total litter C input	0.8974	4.84×10^{-10}	19
Cumulative CO_2 emissions ~ water-soluble litter C input	0.798	1.606×10^{-7}	19
Litter-derived CO_2 flux ~ SOM-derived CO_2 flux	0.8838	$<2.2 \times 10^{-16}$	495
$\delta^{15}N_{NO3}$ of soil NO_3^- ~ net N mineralization (52 DAO)	0.729	$<2.024 \times 10^{-6}$	19

3.3. CO_2 and $^{13}CO_2$ Fluxes and Cumulative Emissions

CO_2 fluxes from all litter treatments increased after onset of incubation compared to Control (Figure 1a,b, Supplementary Figure S3). Total CO_2 fluxes were highest in Leaf reaching 5.1 mg C kg^{-1} h^{-1} on 2 DAO. In Root, CO_2 flux peaked on 2 DAO (2.57 mg C kg^{-1} h^{-1}) and then decreased throughout the incubation period. In Straw, the highest CO_2 fluxes were measured directly after onset of incubation (2.8 mg C kg^{-1} h^{-1}), and continuously decreased afterwards. Litter-derived and SOM-derived CO_2 followed a similar pattern as total CO_2 fluxes and were highly correlated (adj. R^2 = 0.88, *p* < 0.001, Table 3). Highest litter-derived CO_2 fluxes were measured in Leaf on 3 DAO (3.0 mg C kg^{-1} h^{-1}), in Straw on 1 DAO (1.2 mg C kg^{-1} h^{-1}), and in Root on 2 DAO (1.7 mg C kg^{-1} h^{-1}) (Figure 1a). SOM-derived CO_2 was highest in Leaf and higher in all litter treatments compared to Control for the first week after onset of incubation (Figure 1b). Accordingly, the cumulative priming effect increased most strongly in all litter treatments during the first days of incubation (Figure 1c) with highest values in Leaf.

Cumulative CO_2 emissions from all litter treatments were significantly higher than from Control without litter (*p* < 0.05, Table 4). The highest cumulative and litter-derived CO_2 emissions were measured after addition of maize leaves, followed by maize straw and maize roots; however, cumulative SOM-derived CO_2 emissions were higher than litter-derived CO_2 emissions in all treatments (Table 4). Total cumulative CO_2 emissions were significantly positively correlated with total C and water-soluble C input from maize litter (adj. R^2 = 0.80 and adj. R^2 = 0.90, respectively, *p* < 0.001, Table 3). When total cumulative CO_2 emissions were standardized against C input from litter, no differences were found (Supplementary Table S2).
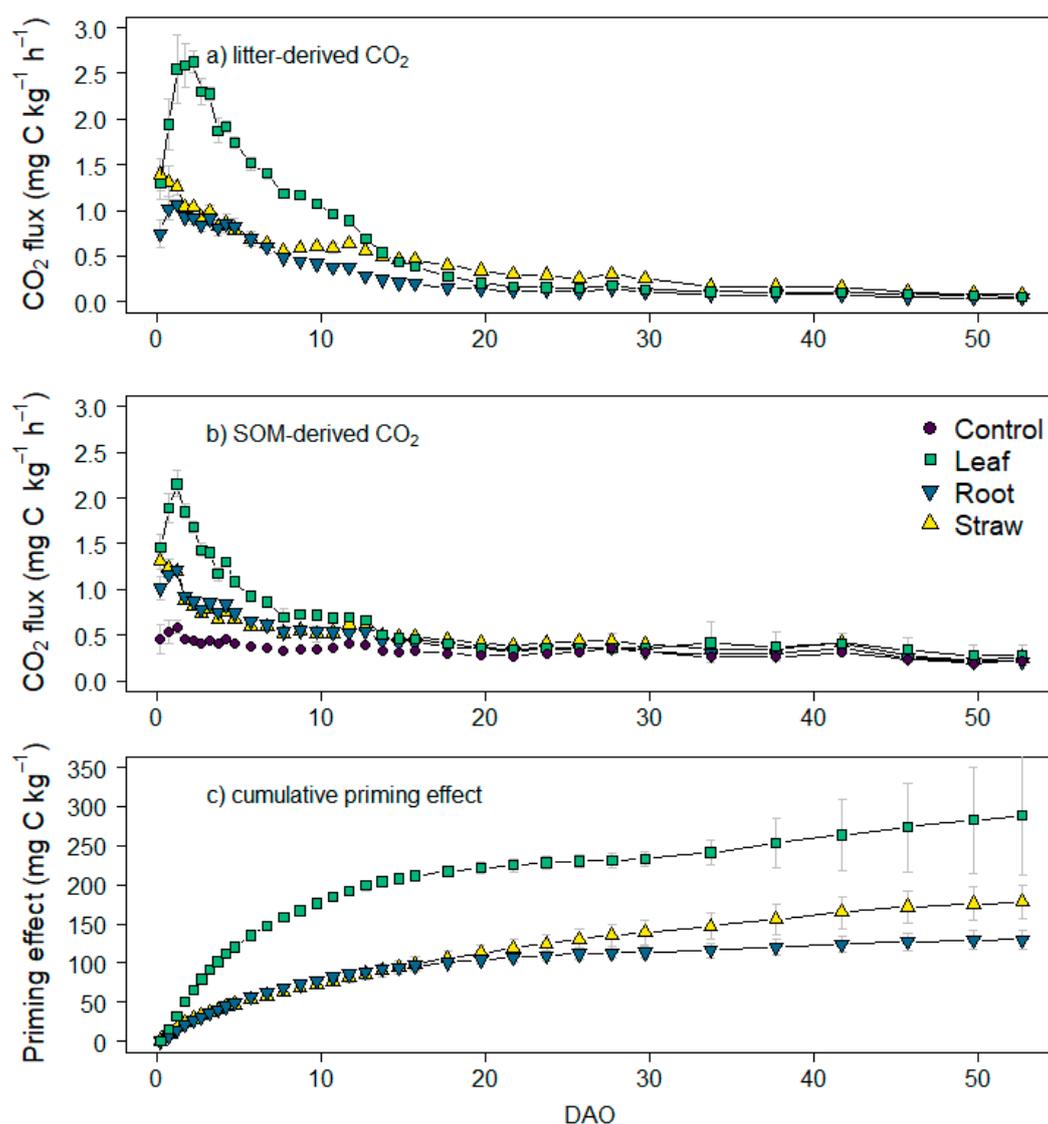
Figure 1. (**a**) Litter-derived CO_2 fluxes, (**b**) SOM-derived CO_2 fluxes, and (**c**) cumulative priming effect (means and standard deviation for n = 5, n = 4 for Control, when not visible, error bars are smaller than the symbols).

Table 4. Cumulative SOM and litter-derived CO_2 emissions, and priming effect.

	Total CO_2 (mg C kg^{-1} dry soil)		SOM-Derived CO_2 (mg C kg^{-1} dry soil)		Litter-Derived CO_2 (mg C kg^{-1} dry soil)		Priming Effect (mg C kg^{-1} dry soil)	
Control	359.5 ± 13.2	d	359.5 ± 13.2	c	-		-	
Maize Leaves	1266.0 ± 118.8	a	654.8 ± 83.5	a	597.5 ± 33.9	a	288.2 ± 76.2	a
Maize Roots	749.8 ± 68.1	c	504.9 ± 10.7	b	281.6 ± 17.6	c	130.0 ± 12.0	b
Maize Straw	970.8 ± 34.3	b	561.9 ± 26.9	b	449.7 ± 21.1	b	178.4 ± 21.5	b

Values represent means (n = 5, for Control n = 4) $\pm$ standard deviation. Different letters in the same column indicate a significant difference.

3.4. N Fluxes and Cumulative Emissions

During the oxic incubation phase, only N_2O and NO fluxes were measured as N_2 fluxes were below the detection limit (Figure 2a,b). N_2O fluxes from litter treatments were higher than 3.7 µg N kg^{-1} h^{-1} for the first measurements on 1 DAO and declined to <1 µg N kg^{-1} h^{-1} until 5 DAO. N_2O fluxes from litter-amended soils were in tendency higher than N_2O fluxes from Control. Initial NO fluxes were ~0.08 µg N kg^{-1} h^{-1} in Control and Root, and ~0.06 µg N kg^{-1} h^{-1} in leaves and straw. In Leaf, a second NO

peak was detected on 5 DAO. NO fluxes in Control were in tendency higher than in all litter treatments until 14 DAO, while NO flux declined fastest in Straw, where fluxes were smaller than in all other treatments after 3 DAO. The ratio of NO/N_2O was highest in Control directly after onset of incubation with maximum values of 0.47 (Supplementary Figure S5). In Root and Leaf, it reached maximum values of 0.2 and 0.1 on 5 and 6 DAO. In Straw, highest measured values were 0.1 on 2 DAO. With onset of the anoxic phase, NO/N_2O decreased to 0.015 in all treatments.
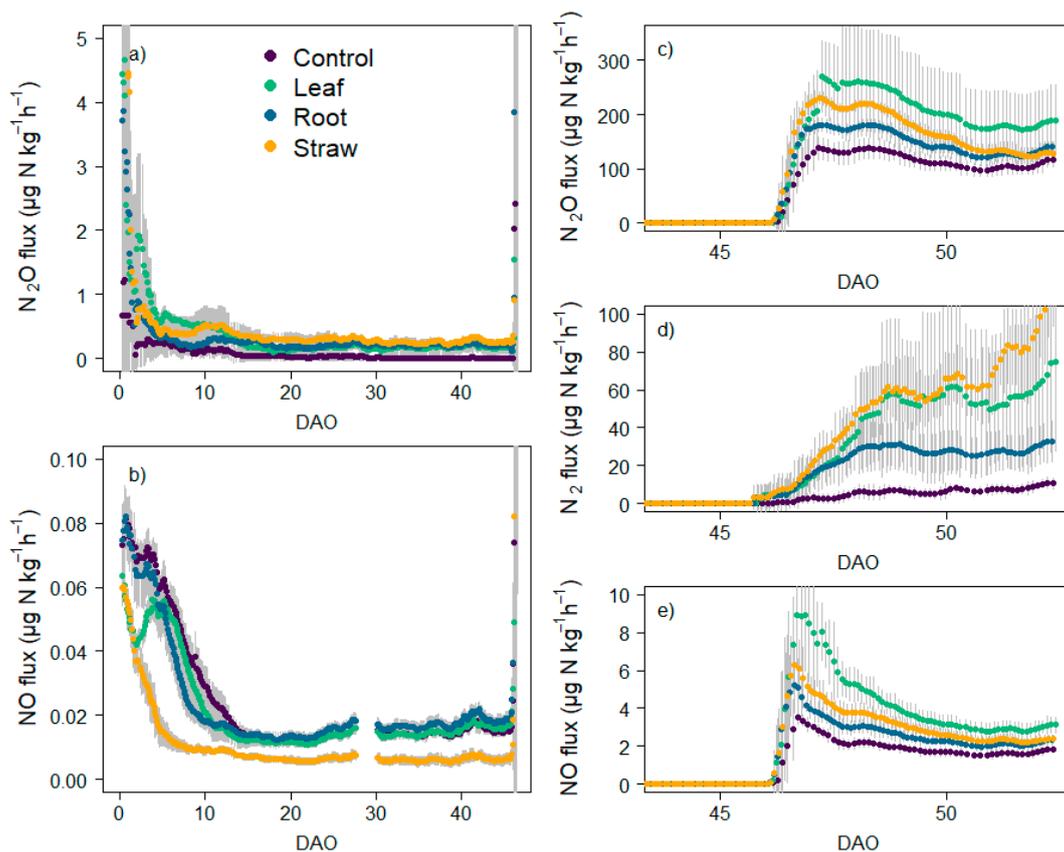
Figure 2. (**a**,**b**) N_2O, and NO fluxes during oxic incubation period (0–47 DAO). (**c–e**) N_2O, N_2, and NO fluxes during anoxic incubation (47–55 DAO) of maize litter on grassland soil (means and standard deviation for n = 5, n = 4 for Control, when not visible, error bars are smaller than the symbols).

During the oxic phase, cumulative N_2O emissions from litter-amended soil were higher than from Control ($p < 0.05$, Table 5). Cumulative emissions in Straw were higher than in Root and similar to Leaf. Cumulative NO emissions were highest in Control and lowest in Straw, and NO/N_2O ratio was significantly higher in Control than in litter-amended treatments. NO emissions strongly decreased with decreasing litter C:N ratio (adj. $R^2 = 0.86$, $p < 0.001$) and increased with increasing N mineralization (adj. $R^2 = 0.57$, $p < 0.001$) confirming that litter quality affected nitrification-derived NO emissions during the oxic incubation phase.

Table 5. Cumulative NO, N_2O, and N_2 emissions and ratios of gaseous products during oxic and anoxic incubation.

	Oxic Incubation Phase (0–46 DAO)								Anoxic Incubation Phase (47–55 DAO)								
	Cumulative NO (μg N kg^{-1} dry soil)		Cumulative N_2O (μg N kg^{-1} dry soil)		NO/N_2O		Cumulative NO (mg N kg^{-1} dry soil)		Cumulative N_2O (mg N kg^{-1} dry soil)		Cumulative N_2 (mg N kg^{-1} dry soil)		$N_2O/(N_2O+N_2)$	
Control	24.1 ± 2.5	a	78.3 ± 97.2	c	0.37 ± 0.19	a	0.29 ± 0.04	c	16.6 ± 2.5	c	0.88 ± 0.33	c	0.95 ± 0.03	a
Maize Leaves	20.8 ± 1.4	b	387.2 ± 94.4	ab	0.05 ± 0.02	b	0.64 ± 0.10	a	29.8 ± 9.0	a	6.75 ± 4.28	ab	0.83 ± 0.04	b
Maize Roots	22.9 ± 2.8	ab	319.0 ± 81.0	b	0.07 ± 0.01	b	0.41 ± 0.06	b	21.5 ± 1.5	bc	3.70 ± 1.58	bc	0.85 ± 0.05	b
Maize Straw	10.0 ± 1.7	c	552.2 ± 260.7	a	0.02 ± 0.01	b	0.48 ± 0.06	b	24.8 ± 1.0	ab	8.36 ± 2.06	ab	0.75 ± 0.05	c

Values represent means (n = 5, for Control n = 4) $\pm$ standard deviation. Different letters in the same column indicate a significant difference according to the LSD post hoc test at $p \leq 0.05$. N.b. different units for gas emissions during oxic and anoxic incubation phases.

After 47 days, anoxic incubation conditions were induced by flushing the headspace with pure helium gas. N_2O, NO, and N_2 fluxes strongly increased with onset of anoxic incubation conditions (Figure 2c–e). N_2O and NO fluxes peaked on 48 DAO and then decreased until the end of the experiment. N_2 fluxes increased after onset of anoxic conditions until the end of the experiment. During the anoxic phase, cumulative N_2O, NO, and N_2 emissions were higher in litter treatments than in Control, although the effect was not always statistically significant for maize roots (Table 5). The highest cumulative emissions were measured for NO (0.64 mg N kg^{-1}) and N_2O in Leaf (29.8 mg N kg^{-1}), and for N_2 in Straw (8.4 mg N kg^{-1}). The ratio of the gaseous end products $N_2O/(N_2O+N_2)$ was highest in Control (0.95) and lowest in Straw (0.75).

3.5. N_2O Isotopocule Mapping Approach, fbD and rN$_2$O Values

The $\delta^{15}N^{SP}/\delta^{18}O_{N2O}$ isotopocule map showed a strong influence of the incubation day on the isotopic signature of soil-emitted N_2O (Figure 3). Most data points were distributed between the mixing line of bacterial and fungal denitrification and the N_2O reduction line during the oxic incubation phase (0–47 DAO). With onset of anoxic incubation conditions, bacterial denitrification became the dominant process as samples measured on 48 DAO cluster tightly above the reported ranges for heterotrophic bacterial denitrification. With ongoing anoxic incubation, the samples cluster along the reduction line indicating increasing N_2O reduction with ongoing anoxic incubation conditions (51 and 54 DAO).
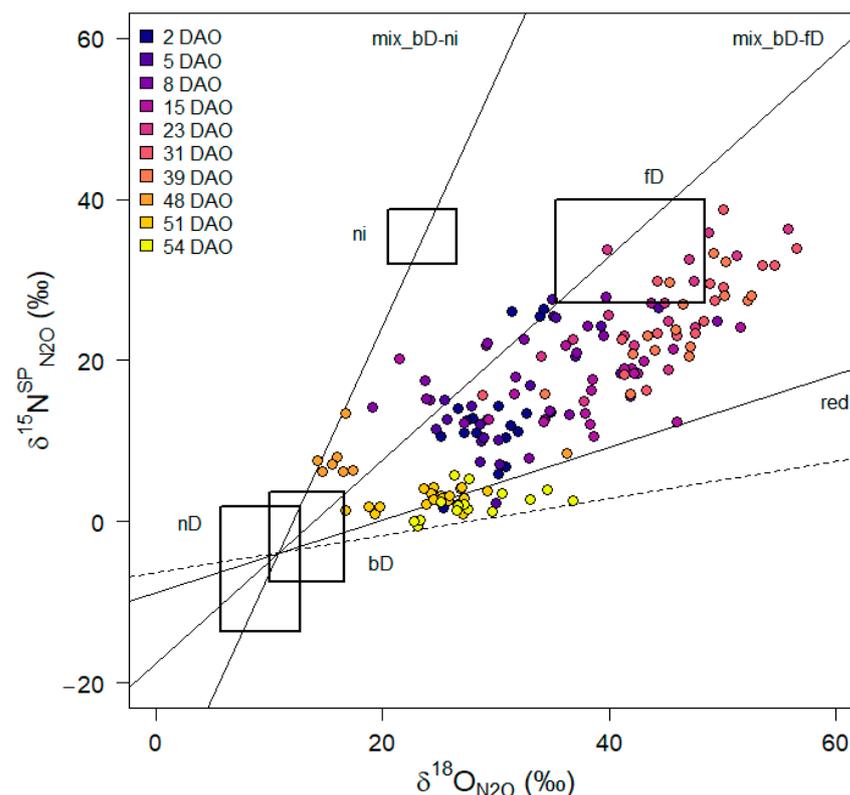
Figure 3. Isotopocule values of soil-emitted N_2O per day plotted in the isotopocule map based on Lewicka-Szczebak et al. (2017) [22] and Yu et al. (2020) [23]. Boxes indicate the mean ranges for end-member values of $\delta^{15}N^{SP}_{N2O}$ and $\delta^{18}O_{N2O}$ (corrected for $\delta^{18}O_{H2O}$) for heterotrophic bacterial denitrification (bD), nitrifier denitrification (nD), nitrification (ni), and fungal denitrification (fD) (view Table S1 for details). The mixing line connects the mean values of bD and fD (mix_bD-fD) or bD and ni (mix_bD-ni), respectively. The slope of the reduction lines (red) is based on the isotopic fractionation factor associated with N_2O reduction to N_2. Dashed line represents the minimum reduction line (n = 178, oxic incubation conditions from 0 DAO to 46 DAO, anoxic incubation conditions from 47 DAO to 55 DAO).

$\delta^{15}N_{bulk}$ and $\delta^{18}O_{N2O}$ of soil-emitted N_2O followed a similar pattern. Both values increased slightly during anoxic incubation (Figures 4 and S6). With onset of anoxic conditions, both $\delta^{15}N_{bulk}$ and $\delta^{18}O_{N2O}$ decreased strongly and increased again until the end of the experiment.
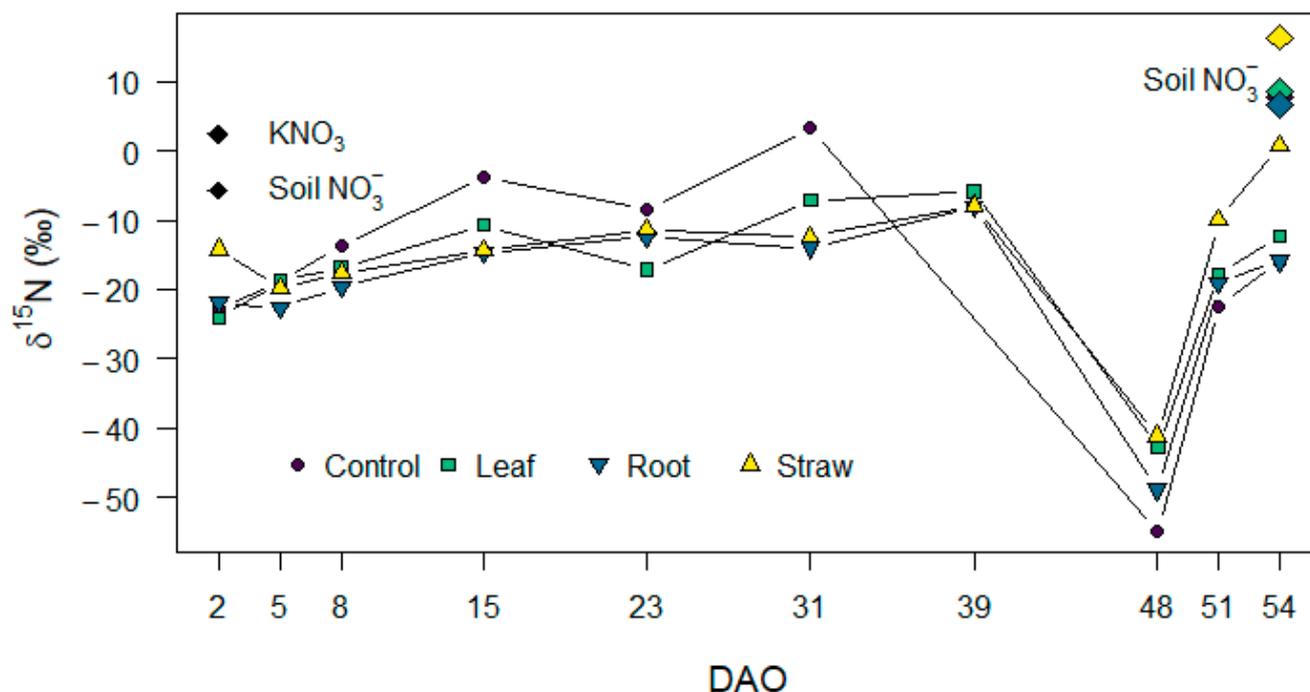
Figure 4. $\delta^{15}N_{bulk}$ of N_2O (colored symbols and lines), added KNO_3 and soil NO_3^- at first day of incubation (black symbols) and soil NO_3^- at last day of incubation (colored symbols with black borders) (means for n = 5, n = 4 for Control).

fbD and rN_2O values exhibited similar patterns for mixing between heterotrophic bacterial denitrification/nitrifier denitrification and nitrification or fungal denitrification (Figure 5a,b, Supplementary Figure S6). After onset of incubation, the fraction of soil-emitted N_2O from heterotrophic bacterial denitrification/nitrifier denitrification (fbD, Figure 5a) was in tendency higher in maize litter treatments compared to Control. While fbD decreased in maize litter treatments during the oxic incubation period, it increased in Control without litter addition. With onset of anoxic incubation conditions, fbD increased strongly in all treatments, reaching values > 0.9, indicating that bacterial denitrification became the dominant process under anoxic incubation conditions. The residual, unreduced N_2O fraction (rN_2O, Figure 5b) was mostly < 0.5 and decreased during the oxic incubation phase, highlighting the significance of N_2O reduction. On 51 DAO, rN_2O was highest and decreased until 54 DAO.
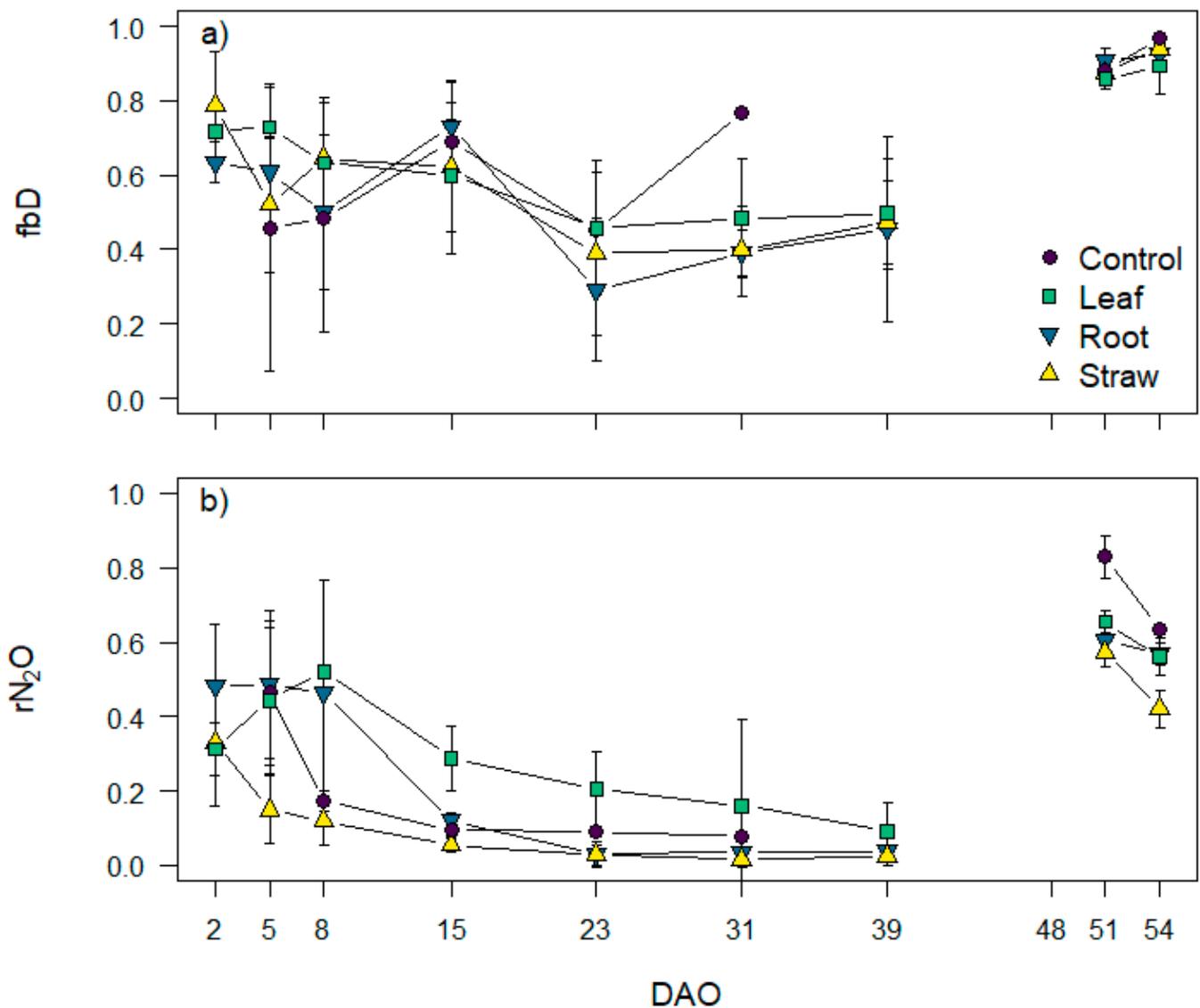
Figure 5. (**a**) Fraction of N_2O originating from heterotrophic bacterial denitrification/nitrifier denitrification (fbD) and (**b**) fraction of residual unreduced N_2O (rN_2O). Values were calculated based on the isotopocule mapping approach by Lewicka-Szczebak et al. (2017) [22] and represent results for scenario 1 (reduction-mixing) of bacterial denitrification with fungal denitrification (mean and standard deviation for n = 5, n = 4 for Control, data points missing for samples with an isotopic signature outside the reduction-mixing area, no N_2O emitted from Control on 39 DAO).

3.6. Interactions between C and N Availability and N Fluxes

To test the effect of C availability and SOM turnover on N fluxes and cumulative emissions, simple linear regression models were tested (Table 3).

The relationship between hourly $NO+N_2O$ and CO_2 fluxes was very weak during the oxic incubation phase (adj. $R^2 = 0.08$, $p < 0.001$, Table 3). In contrast, $NO+N_2O+N_2$ fluxes were highly positively correlated with CO_2 fluxes during the anoxic incubation period (adj. $R^2 = 0.86$, $p < 0.001$, Table 3). Similarly, fbD and rN_2O were positively correlated with total CO_2 flux from soil, but the relationship was weak (adj. $R^2 = 0.23$ and 0.31, respectively, $p < 0.001$, Table 3).

Cumulative anoxic N emissions ($NO+N_2O+N_2$) were significantly positively correlated with total litter C input (adj. $R^2 = 0.51$, $p < 0.001$, Table 3). Furthermore, we found a significant negative correlation between the ratio of cumulative $N_2O/(N_2O+N_2)$ emissions during anoxic incubation and the ratio of water-soluble C:N of maize litter types (adj.

$R^2 = 0.51$, $p < 0.001$, Table 3) and also with the soil NO_3^-:WEOC ratio at the end of the experiment (adj. $R^2 = 0.41$, $p < 0.01$, Table 3). When standardized against C input from litter, total cumulative N emissions did not differ between litter treatments (Supplementary Table S2).

4. Discussion

4.1. Maize Litter Quality Controls N Mineralization

Soil mineral N content, mineralization, and immobilization strongly depended on maize litter quality. In Control without litter addition, soil NH_4^+ content decreased, but soil NO_3^- content strongly increased due to mineralization and nitrification of soil organic N. Tillage often leads to increased soil mineral N content [40,41], and net N mineralization from control soils without litter addition has been reported [8,13,42]. In Leaf and Root, mineralization was higher than in Control as additional organic N from litter was mineralized. In contrast, addition of maize straw immobilized N, which coincides with the higher C:N ratio in Straw compared to Leaf and Root. In general, immobilization of N follows the addition of litter with C:N ratios > 25:1 [7]. Net N mineralization after addition of maize roots is in contrast to most other studies reporting net N immobilization after addition of maize roots [8,13,42,43]. However, we used maize roots grown in a nutrient solution, which had higher total and water-soluble N content than those reported in other studies [13,44].

4.2. Effect of Maize Litter Quality on CO₂ Emissions and Priming Effect

After onset of incubation, both litter and SOM-derived CO_2 fluxes strongly increased in litter treatments, while CO_2 efflux in Control was stable. Total cumulative CO_2 emissions increased with increasing input of total and water-soluble C, indicating that decomposition dynamics were controlled by the amount and quality of added plant material. The chemical composition of plant litter is known to be a primary controller of decomposition rates of both roots [11,12,15] and aboveground plant litter [4,10].

Especially during the first week after litter addition, SOM turnover was increased in all litter treatments, leading to a positive priming effect. The highest increase in SOM turnover was observed after addition of maize leaves which were characterized by high amounts of easily degradable compounds as indicated by high shares of carboxyl C (i.e., organic acids, amino acids) and alkyl C (i.e., fatty acids, amino acids, paraffines) [45]. Litter and SOM-derived CO_2 fluxes followed the same pattern and were highly positively correlated confirming that litter and SOM turnover are interrelated. Thus, our results are in accordance with the concept that litter with high degradability increases SOM turnover leading to a positive priming effect [17–19].

4.3. Effect of Litter Quality and N Mineralization on N Emissions and Production Pathways under Oxic Atmosphere

Directly after onset of oxic incubation, NO fluxes were highest in Control and Root while N_2O fluxes were higher in litter amended treatments than in Control. The ratio of NO/N_2O can be used as an indicator whether NO is produced from nitrification or denitrification [46,47]. While the NO/N_2O emission ratio of bacterial denitrification is mostly around 0.01, the emission ratios of NO/N_2O from nitrification are often higher than 1 [47]. In our study, the emission ratio of NO/N_2O was highest during the first 10 days after onset of incubation, with maximum values of 0.47 in Control indicating a high contribution of nitrification to NO formation. Analysis of soil samples taken prior to the onset of incubation revealed high NH_4^+ content of soil, which further supports that nitrification contributed to NO emissions during the initial incubation phase. In Straw, where N was immobilized to decompose C compounds, NO fluxes decreased faster and were lower than in all other treatments. Furthermore, oxic cumulative NO emissions strongly decreased with increasing litter C:N ratio (adj. $R^2 = 0.86$, $p < 0.001$) and increased with increasing N mineralization (adj. $R^2 = 0.57$, $p < 0.001$) confirming that litter quality affected nitrification-derived NO emissions in the beginning of the oxic incubation phase.

Addition of maize litter increased N_2O fluxes compared to non-amended Control. As first, the headspace atmosphere had to be replaced by the mixture of He/O_2, measurements started approximately one day after onset of incubation conditions. At this time, decreasing N_2O fluxes indicated that N_2O fluxes had peaked within 24 h after water and NO_3^- addition. After this initial increase, N_2O fluxes decreased rapidly and then stayed on a similar level throughout the oxic phase. Immediately after onset of the incubation, bacterial denitrification (i.e., heterotrophic bacterial denitrification and nitrifier denitrification) was the dominant N_2O emitting process in litter-amended treatments as indicated by the fraction of bacterial denitrification (fbD) > 0.6. Gradually decreasing fbD values then indicate a shift towards nitrification or fungal denitrification. Litter addition [27,48] and soil moisture of 70% WFPS may have promoted fungi, which often contribute to denitrification under weakly anoxic conditions [49,50]. Several studies have described a shift from bacterial to fungal dominance with ongoing incubations [27,51–53]. However, nitrification may have contributed to N_2O formation in Root and Leaf as indicated by high mineralization and NO/N_2O ratio. In Control, nitrification presumably contributed to N_2O formation, especially during the first days of the experiment, when fbD was < 0.3 and the NO/N_2O ratio was high.

rN_2O values were mostly < 0.5, highlighting the significance of N_2O reduction, also during the oxic incubation period. Thus, although N_2 fluxes were lower than the detection limit of our incubation system (5.5 µg N_2-N kg^{-1} h^{-1}), they significantly contributed to gaseous N losses. N_2O reduction to N_2 is the last step of denitrification [54] and it usually takes place when availability of NO_x is limited [55]. Furthermore, pore size and distribution, and soil moisture may affect N_2O reduction to N_2, as they control diffusion of O_2 and N_2O in soil [20,56]. Accordingly, for interpretation of the isotopocule mapping approach in our experiment, we think that the reduction-mixing scenario is more plausible: N_2O was produced by denitrifying bacteria and partly reduced to N_2 in anoxic microsites, and then N_2O diffusing out of these hotspots was mixed with N_2O from nitrification and fungal denitrification [57]. We anticipate that nitrification contributed to N_2O formation when mineralization was high, while fungal denitrification became more important in litter-amended treatments with ongoing incubation.

4.4. Effect of Maize Litter Quality and Mineralization on Potential Denitrification

With onset of anoxic incubation conditions on 47 DAO, total NO and N_2O fluxes increased rapidly, while N_2 fluxes increased more slowly. $\delta^{18}O_{N2O}$ of N_2O emitted on 48 DAO falls in the range of heterotrophic bacterial denitrification reported in earlier studies [23], indicating that heterotrophic bacterial denitrification was the main N_2O-emitting process with low reduction to N_2. Interestingly, the $\delta^{15}N^{bulk}$ values on 48 DAO strongly deviated from measured $\delta^{15}N^{bulk}$ values on all other sampling days and were slightly outside the reported endmember values of heterotrophic bacterial denitrification. Under oxic conditions, denitrification mostly took place in anoxic hotspots where ongoing reduction led to a fractionation in the NO_3^- pool undergoing denitrification, which is reflected in gradually increasing $\delta^{15}N^{bulk}$ values. With onset of anoxic conditions, previously unreduced NO_3^- contributed to N_2O formation leading to a shift towards more negative $\delta^{15}N^{bulk}$ values [29,58]. When the contribution of pools with different N dynamics changes, shifts in the isotopic signature have been reported [59–61]. For our study, low $\delta^{15}N^{bulk}$ values on 48 DAO are consistent with very high N_2O fluxes and the low measured $N_2O/(N_2O + N_2)$ ratio on 48 DAO leading to strong fractionation effects.

Analysis of $\delta^{15}N$ and $\delta^{18}O$ in soil NO_3^- may improve accuracy of the N_2O mapping approach and estimation of N_2O formation processes [23]. $\delta^{15}N_{NO3}$ was higher at the end of the experiment compared to initial soil NO_3^- and added KNO_3, confirming the ongoing fractionation during the reduction of the soil NO_3^- pool. Furthermore, $\delta^{15}N_{NO3}$ increased with decreasing mineralization, indicating a higher share of added KNO_3 to residual NO_3^- at the end of the incubation experiment. Higher $\delta^{18}O$ in Straw and Root may point towards a higher contribution of fungal denitrification, which is consistent with

the higher contribution of fungi to decomposition of plant materials rich in celluloses and lignin [16,27]. Overall, it needs to be taken into account, that estimating N_2O formation processes based on N_2O isotopomers is subject to large uncertainties. Endmember values, reduction and fractionation factors have been obtained under differing incubation and environmental conditions, and may thus lead to over or underestimation of contributing processes [57,61,62].

4.5. Interaction between C Turnover and Denitrification

In agricultural soils, denitrification is often controlled by the availability of readily decomposable organic matter with increasing C availability leading to increased N losses [6,13,14,63,64]. In contrast, we measured low denitrification derived N fluxes under oxic conditions, and the correlation between oxic N and CO_2 fluxes was very weak indicating that denitrification was not directly affected by litter decomposition in our study. Although soil NO_3^- content was high and high litter and SOM turnover confirmed high C_{org} availability, N_2O fluxes were very low, indicating that conditions for denitrifying microorganisms were not optimal. Rohe et al. (2021) [65] reported very low $N_2O + N_2$ fluxes from an incubation study with the same soil at 60% WFPS and higher fluxes compared to our study for 75 and 85% WFPS. Thus, our incubation conditions with a soil moisture of 70% WFPS may have been too low to promote denitrifying soil microorganisms. With onset of anoxic conditions, N fluxes increased immediately, confirming that high pO_2 was restricting denitrification during the oxic incubation phase. In contrast to our expectations, the microbial respiration of litter and SOM did not promote the formation of litter associated anoxic hotspots for denitrification as high O_2 diffusivity limited denitrification [65,66].

In contrast, N and CO_2 fluxes were highly positively correlated (adj. $R^2 = 0.86$, $p < 0.001$) under anoxic conditions, and cumulative N emissions increased with increasing litter C input (adj. $R^2 = 0.51$, $p < 0.001$) confirming our hypothesis that higher C availability leads to increased gaseous N losses. However, this effect was based on the role of C as energy source for denitrifiers, as the potential O_2 consumption by C decomposition was not relevant under anoxic conditions.

The ratio of denitrification end products $N_2O/(N_2O+N_2)$ decreased with increasing water-extractable C:N ratio of litter (adj. $R^2 = 0.73$, $p < 0.001$) and increasing soil $WEOC:NO_3^-$ ratio at the end of the experiment (adj. $R^2 = 0.41$, $p < 0.01$) confirming that the ratio of available C to oxidized N is a major control of denitrification product stoichiometry [67]. Immobilization of N after addition of maize straw with high C:N ratio restricted N availability leading to higher N_2O reduction to N_2. However, as soil NO_3^- content was still very high (> 70 mg NO_3^--N kg^{-1} in all treatments at the end of the incubation experiment), N_2O was the dominant end product, as NO_3^- is preferentially used as electron acceptor and high soil NO_3^- content can inhibit the reduction of N_2O to N_2 [27,67].

5. Conclusions

We investigated the effect of different maize litter types (young leaves and roots, straw) on CO_2, NO, N_2O, and N_2 emissions under oxic and anoxic conditions in a laboratory incubation study. Addition of maize litter increased litter and SOM derived CO_2 emissions, leading to a positive priming effect. SOM priming was highest after addition of maize leaves with a high share of easily degradable C compounds during the first week after onset of incubation. Although litter and SOM turnover were high, NO and N_2O fluxes were low under oxic conditions as high O_2 diffusivity limited denitrification.

The NO/N_2O ratio indicated that nitrification contributed to NO and N_2O formation during the first two weeks of incubation, especially in Control without litter addition. In the litter-amended treatments, isotopocule mapping revealed that bacterial denitrification dominated N_2O formation in the beginning of the incubation experiment with a subsequent shift towards fungal denitrification. With onset of anoxic incubation conditions after 47 days, N fluxes strongly increased and heterotrophic bacterial denitrification became the dominating source of N_2O. The $N_2O/(N_2O + N_2)$ ratio decreased with increasing litter

C:N ratio and C_{org}:NO_3^- ratio in soil confirming that the ratio of available C:N is a major control of denitrification product stoichiometry.

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