Synthesis and X-Ray Crystal Structure of Germaimines

Anton Meller,* Günter Ossig, Walter Maringgele, Dietmar Stalke, Regine Herbst-Irmer, Stefanie Freitag and George M. Sheldrick

Institut für Anorganische Chemie, Universität Göttingen, Tammannstraße 4, D-3400 Göttingen, Germany

Three new germaimines have been synthesized and two of them characterized by X-ray diffraction; the Ge=N bond lengths agree well with an *ab initio* prediction for H₂Ge=NH.

The Ge=N distance in unsubstituted germaimine (H₂Ge=NH) has been predicted to be 169.5 pm by *ab initio* calculation with double- ζ (DZ) plus polarization basis sets.¹ Germaimines have been observed as transient species (and trapped by various substrates) since 1978 (see ref. 2 for a recent review). Only a few examples have been isolated and spectroscopically characterized within the last few years.^{3–8} Here we report the synthesis of three new germaimines prepared from highly sterically hindered diazagermylenes⁹ and 2,6-disubstituted phenylazides as shown in Scheme 1.[†] In the ¹⁵N NMR spectrum of **3** two signals with 2:1 intensity ratio at $\delta = -295$ and -193 (rel. to MeNO₂) have been detected.

Compounds 1–3 crystallized easily from hot THF (tetrahydrofuran) to give well-shaped amber crystals, which change from yellow at -196 °C to red at 150 °C. Compound 1 hydrolyses in air to give [(2,6(Pri)₂C₆H₃(Me₃Si)N]₂Ge(OH)– NHC₆H₃–2,6(Pri)₂ 4 (from spectroscopic evidence). Likewise, methanol is added across the (Ge=N) double bond of 3 thus

† Preparative details and spectroscopic data: To a solution of the appropriate diazagermylene (0.020 mol) {bis[(2,6-diisopropylphenyl)-(trimethylsilyl)amino]germane(II) for 1, bis[(2,4,6-trimethylphenyl)(trimethylsilyl)amino]germane(II) for 2 and 3} in 80 ml THF the equimolar amount of arylazide (2,6-diisopropylphenylazide for 1 and 2, 2,4,6-trimethylphenylazide for 3) dissolved in 40 ml THF was added dropwise, with stirring, at ambient temperature and maintained at 25 °C for 2 h. The clear, deep-red reaction mixtures was concentrated under reduced pressure, the amber crystals formed were filtered off and dried *in vacuo*. Satisfactory C,H,N elemental analyses were obtained for 1–3 and 5.

1: Bis[(2,6-diisopropylphenyl)(trimethylsilyl)amino]-(2,6-diisopropylphenylimino)germane: m.p. 195 °C (decomp.); yield 11.0 g (74%); C₄₂H₆₉GeN₃Si₂ (744.79); MS [*m*/*z* (%)]: EI 745 (20) [M⁺], 73 (100); NMR (in C₆D₆): δ (¹H) 0.33 (s, 18H, SiMe₃); 1.06, 1.28, 1.32 (3 × d, ³J_{HH} 6.6 Hz, 3 × 6H, 6 × CH₃-Prⁱ); 1.40–1.60 (br, 18H, 6 × CH₃-Prⁱ); 2.90, 3.80, 4.06 (3 × sept., 3 × 2H, 6 × CH-Prⁱ); 6.70–7.10, 7.24–7.32 (2 × br, 9H, arom.); δ (¹3C) 3.57 (SiMe₃); 24.63, 24.73, 25.36, 25.68 (threefold intensity) (all CH₃-Prⁱ); 28.71, 29.06, 29.15 (all CH-Prⁱ); 120.05, 123.25, 124.93, 125.47, 127.08, 139.30, 141.67, 147.15, 147.52, 148.65 (10 signals apparently due to non-equivalence of the 2/6 and 3/5 positions of the imino-bonded aromatic ring) (all arom.); δ (²9Si) 16.2.

arom,), 6(~3) 10.2. **2**: Bis[(2,4,6-trimethylphenyl)(trimethylsilyl)amino]-(2,6-diisopropylphenylimino)germane: m.p. 162 °C; yield 9.2 g (70%); C₃₆H₅₇GeN₃Si₂ (660.63); MS [*m*/*z* (%)]: EI 661 (22) [M⁺], 351 (100); NMR (in C₆D₆): δ (¹H) 0.08 (s, 18H, SiMe₃); 1.50 (d, ³*J*_{HH} 6.8 Hz, 12H, 4 × CH₃-Prⁱ); 1.97 (s, 12H, 4 × o-CH₃); 2.08 (s, 6H, 2 × p-CH₃), 3.86 (sept., 2H, 2 × CH-Prⁱ); 6.46–6.52 (m, 4H, C₆H₂); 7.05–7.13, 7.28–7.34 (2 × m, 3H, C₆H₃); δ (¹³C) 2.12 (SiMe₃); 20.11 (o-CH₃); 20.72 (*p*-CH₃); 23.98 (CH₃-Prⁱ); 29.38 (CH-Prⁱ); 119.92, 122.63, 129.37, 134.00, 136.42, 139.20, 140.58, 149.54 (all arom.); δ (²⁹Si) 14.1.

3: Bis[(2,4,6-trimethylphenyl)(trimethylsilyl)amino]-(2,4,6-trimethylphenylimino)germane: m.p. 177 °C; yield 8.54 g (69%); $C_{33}H_{51}GeN_3Si_2$ (618.55); MS [m/z (%)]: EI 619 (21) [M+], 351 (100); NMR (in C₆D₆): δ ('H) 0.11 (s, 18H, SiMe_3); 2.03 (s, 18H, o-CH₃ of mes + mes'); 2.33 (s, 3H, p-CH₃ of mes'/imino); 2.61 (s, 6H, p-CH₃ of mes/amino); 6.380, 6.382 (2s, 4H, C₆H₂ of mes); 7.064, 7.066 (2s, 2H, C₆H₂ of mes); δ (¹³C) 1.95 (SiMe₃); 20.66 (o-CH₃, mes/amino); 21.04 (p-CH₃, mes'); 127.30, 128.21, 129.09, 129.25, 133.63, 135.66, 140.61, 150.11, (C₆H₂ mes + mes'); δ (²⁹Si) 13.4.— δ (¹⁵N)-NMR (40% sol. in THF, standard CH₃NO₂ ext.): -295 (2N), -193 (1N).

 $[R(Me_{3}Si)N]_{2}Ge + N_{3}R' \xrightarrow{\text{THF}, 20 \,^{\circ}\text{C}} [R(Me_{3}Si)N]_{2}Ge=NR'$ $1: R = R' = 2,6(Pr^{i})_{2}C_{6}H_{3}$ $2: R = 2,4,6-Me_{3}C_{6}H_{2}, R' = 2,6(Pr^{i})_{2}C_{6}H_{3}$ $3: R = R' = 2,4,6-Me_{3}C_{6}H_{2}$ $3 + MeOH \xrightarrow{\text{THF}, 20 \,^{\circ}\text{C}} [R(Me_{3}Si)N]_{2}Ge(OMe)-NHR \cdot \text{THF}$

 $5 (R = 2,4,6-Me_3C_6H_2)$

Scheme 1

forming 5 which crystallizes with 1 mol of THF.‡ The structures of 1 and 3 have been determined by single crystal X-ray diffraction.§ In these molecules, which are stable for months under dry nitrogen at room temperature, the (Ge=N) bond lengths are found to be 170.3(2) pm and 169.1(3) pm, which are in excellent agreement with the calculated value for germaimine.¹ The C-N=Ge-N torsion angles [1.5(5), 178.4(3)° 1 and -1.5(5), 178.1(3)° 3] are consistent with a

NMR (in C_6D_6): $\delta(^{1}H) 0.30$ (s, 18H, SiMe₃); 1.41 (m, 4H, H-3 and H-4 of THF); 1.95, 2.06, 2.14, 2.18 (4s, 27H, *o*-CH₃+*p*-CH₃); 2.91 (s, 1H, NH); 3.56 (m, 4H, H-2 and H-5 of THF); 3.91 (s, 3H, OMe); 6.60–6.80 (br, 6H, C_6H_2); IR (in Nujol) 3350 (vNH).

§ Crystal data for 1: C₄₂H₆₉GeN₃Si₂, triclinic, space group $P\overline{1}$, a = 1144.6(2), b = 1178.7(2), c = 1728.6(3) pm, $\alpha = 88.84(1)$, $\beta = 80.67(1), \gamma = 68.20(1)^\circ, U = 2.13 \text{ nm}^3, Z = 2, D_c = 1.16 \text{ g cm}^{-3}, \mu = 0.80 \text{ mm}^{-1}, F(000) = 804, \text{ crystal dimensions } 0.5 \times 0.5 \times 0.6 \text{ mm},$ 8471 reflections collected with $8 < 2\theta < 50^{\circ}$, 7488 unique and 6099 with $F > 3\sigma(F)$ used in the structural analysis. 3: $C_{33}H_{51}GeN_3Si_2$, monoclinic, space group $P2_1/n$, a = 852.3(2), b = 2961.1(4), c = 1382.1(2) pm, $\beta = 100.30(2)^\circ$, U = 3.43 nm³, Z = 4, $D_c = 1.20$ g cm⁻³, $\mu = 0.99 \text{ mm}^{-1}$, F(000) = 1320, crystal dimensions $0.4 \times 0.4 \times 0.6$ mm, 7086 reflections collected with $8 < 2\theta < 50^{\circ}$, 5991 unique and 4589 with $F > 3\sigma(F)$ used in the structural analysis. Both data sets were collected on a Siemens-Stoe AED diffractometer using Mo-K α radiation ($\lambda = 71.073$ pm) at a temperature of 153 K. Semiempirical absorption corrections were applied. The structures were solved by Patterson and Direct methods. All non-hydrogen atoms were refined anisotropically, and a riding model starting from calculated positions was employed for the hydrogen atoms. 437 1 and 356 3 parameters were refined with a weighting scheme $[w^{-1} = \sigma^2(F) + 0.0002 F_o^2]$ 1 and $[w^{-1} = \sigma^2(F) + 0.00015 F_o^2]$ 3. The refinement converged with R= 0.041, $R_w = 0.037$ 1 and R = 0.048, $R_w = 0.040$ 3 and final difference electron density maxima of 0.51 1 and 0.42 3 and minima of -0.42 1 and -0.41 10^{-6} epm⁻³ 3. Atomic coordinates, bond lengths and angles, and anisotropic displacement parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

[‡] **4** m.p. 236 °C; C₄₂H₇₁GeN₃OSi₂ (762.81); MS [*m*/*z* (%)]: 763 (2), 73 (100); IR (KBr): 3583 (νOH), 3390 (νNH).

^{5:} Bis[2,4,6-trimethylphenyl)(trimethylsilyl)amino]methoxy(2,4,6-trimethylphenylamino)germane was prepared by dropwise addition of a solution of 0.10 g (0.003 mol) MeOH in 5 ml THF to 0.93 g (0.0015 mol) of 3 in 10 ml THF with stirring. After 1 h volatiles were removed *in vacuo* and the colourless residue recrystallized from THF: m.p. 130 °C; yield 0.72 g (67%); C₃₈H₆₃GeN₃O₂Si₂ (722.70); MS [*m*/*z* (%)]: EI 651(8) [M – THF]⁺, 42 (100). NMR (in C₆D₆): $\delta(^{1}$ H) 0.30 (s, 18H, SiMe₃); 1.41 (m, 4H, H-3 and



Fig. 1 Molecular structure of 1; selected bond distances (pm) and angles (°): Ge(1)-N(1) 170.3(2), Ge(1)-N(2) 181.9(3), Ge(1)-N(3) 182.6(3), N(1)-C(1) 141.1(4), N(2)-C(16) 145.3(3), N(3)-C(31)145.5(3); N(1)-Ge(1)-N(2) 108.4(1), N(1)-Ge(1)-N(3) 130.9(1), N(2)-Ge(1)-N(3) 120.7(1)

normal $p\pi$ - $p\pi$ double bond. The difference between the mean Ge-N and Ge=N distances in these compounds is only 12 pm; for most pairs of atoms a double bond is about 20 pm shorter than a single bond. In view of the Ge-N distances reported here, $[GeN(2,6-Pr_{2}^{i}C_{6}H_{3})]_{3}$ [mean Ge-N 185.9(2) pm]¹⁰ should be regarded as a GeII derivative without significant multiple bonding.

Received, 16th May 1991; Com. 1/02326H



Fig. 2 Molecular structure of 3; selected bond distances (pm) and angles (°): Ge(1)-N(3) 169.1(3), Ge(1)-N(1) 180.8(3), Ge(1)-N(2)182.8(3), N(3)-C(25) 139.9(4), N(1)-C(4) 145.2(4), N(2)-C(16)145.7(4); N(3)-Ge(1)-N(1) 110.6(1), N(3)-Ge(1)-N(2) 134.0(1), N(1)-Ge(1)-N(2) 115.4(1)

References

- 1 G. Trinquier, J.-C. Barthelat and J. Satgé, J. Am. Chem. Soc., 1982, 104, 5931.
- 2 J. Satgé, J. Organomet. Chem., 1990, 400, 121
- 3 C. Glidewell, D. Lloyd, K. W. Lumbard and J. S. McKechnie,
- C. Glidewell, D. Lloyd, K. W. Lumbard and J. S. McKechnie, Tetrahedron Lett., 1987, 28, 343.
- 5 J. Pfeiffer, W. Maringgele, M. Noltemeyer and A. Meller, Chem. Ber., 1989, 122, 245.
- 6 H. G. Ang and F. K. Lee, J. Chem. Soc., Chem. Commun., 1989, 310.
- 7 H. G. Ang and F. K. Lee, J. Fluorine Chem., 1989, 43, 435.
- 8 T. Tsumuraya and W. Ando, Chem. Lett., 1989, 1043.
- 9 A. Meller and C. P. Graebe, Chem. Ber., 1985, 118, 2020.
- 10 R. A. Bartlett and P. P. Power, J. Am. Chem. Soc., 1990, 112, 3660.